2005E758

Candidate's Seat No :

M.Sc. (Sem.-III) Examination 504 Chemistry (Org.) May-2017

Time: 3 Hours

|Max. Marks : 70

(Industrial Chemistry)

1.	(a) Discuss the importance of patents in R & D. OR	7
	Discuss the difference between batch versus continuous operation. (b) Write a note on "GLP" and "GMP". OR	7
	Write a note on safety rules of fire and handling of hazardous chemicals.	
2.	(a) Write a note on halogenation. Give importance chlorination for preparation of intermediates.	7
	OR	,
	Discuss sulphonation of Naphthalene, α -Naphthol and β -Naphthol at various concentration of H_2SO_4 and temperature.	
	(b) Various methods for preparation of dyes intermediate by nitration. OR	7
	Discuss amination by ammonolysis of Sulphonation.	
3.	(a) Write a note on O-alkylation and N-alkylation.	*:
	OR	
	Write a brief note on 12 principle of green chemistry.	
	(b) Write a note on green catalysts and solvents with suitable examples. OR	7
	Discuss Michael and Wurtz reaction on the basis of green chemistry.	
4.	(a) Discuss various types of distillation in brief.	7
	OR	
	Write a note on weedicides and fungicides.	
	(b) Discuss various types of filtration.	7
	OR	
	Write a note on plant nutrients and plant hormones.	

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- 5. Answer the following: (Each carry one mark).
 - (i) Define unit process.
 - (ii) What is extraction?
 - (iii) Why grinding is required in industry?
 - (iv) Define agrochemicals.
 - (v) Why iodination is reversible?
 - (vi) Bromination of acetanilide at below 10°C will give which isomer as major product?
 - (vii) What is Oleum?
 - (viii) What is Hydroxylation?
 - (ix) What is ionic liquid?
 - (x) What is aqueous phase reaction?
 - (xi) Which major isomer will form if F.C. acetylation of toluene is carried out by CH₃Cl in presence of AlCl₃?
 - (xii) Define exothermic reaction with one example.
 - (xiii) Define unit operation.
 - (xiv) Which product will form when acetone react with excess of I₂ in presence of NaOH?