

Instruction: Attempt all questions

Q-1

Answer the following

- (a) Explain briefly most probable velocity and derive the equations for the V_{ave} and V_{rms} . (7)

OR

- (a) Derive the equation of Maxwell's law of distribution of molecular speeds. (7)

- (b) Explain how the molecular velocity is calculated on the basis of temperature? Explain the effect of molecular weight of a gas on pressure at specific temperature. (7)

OR

- (b) Derive the kinetic gas equation in terms of kinetic energy. Calculate the average kinetic energy of a H_2 molecule at $0^\circ C$. ($R = 8.314 \times 10^7 \text{ erg K}^{-1} \text{ mol}^{-1}$) (7)

Q-2

Answer the following

- (a) What is the selection rule for vibrational transition in a simple harmonic oscillator in terms of quantum number? Explain briefly the principle of IR spectroscopy. (7)

OR

- (a) Discuss the technique of FT-IR spectroscopy. (7)

- (b) How the force constant is related to the bond order? Explain the selection rules for all transitions in an anharmonic oscillator. (7)

OR

- (b) What is the Hooke's law of IR spectroscopy? Derive the equation for evaluating the bond length of a simple molecule using IR spectroscopy. (7)

Q-3

Answer the following

- (a) What is the basic principle of Raman spectroscopy? Discuss the quantum mechanical treatment of Raman effect. (7)

OR

- (a) Explain the classical theory of Raman scattering. (7)
(b) Discuss the selection rules and some applications of Raman spectroscopy. (7)

OR

- (b) What are the characteristics of Raman lines? What is the fundamental requirement of substance to be Raman active? Explain pure rotational Raman spectra. (7)

Q-4

Answer the following

- (a) Discuss the band theory and some properties of superconductors. (7)

OR

- (a) Explain the BCS theory superconductors. (7)
(b) Explain the mechanical and electrical properties of solid (7)

OR

Q-5

- (b) Discuss the ferro magnetism and anti ferromagnetism. (7)
Answer the following (Any Seven-Two marks each) 14

- (i) What is the effect of temperature on root mean square velocity?
(ii) Why the relative decrease in pressure is larger for a gas having high molecular weight at a particular temperature?
(iii) Calculate the probability density at 500 ms^{-1} for the x-component of the velocity of N_2 at 21.0°C .
(iv) How is bond strength determined?
(v) Calculate the kinetic energy of two molecules of N_2 at 25°C .
(vi) How the conductivity of metal changes with temperature?
(vii) What is the effect of on ferromagnetic substances at Curie temperature?
(viii) What is the difference between stokes and anti-stoke in Raman spectroscopy?
(ix) Explain in brief Raman spectra.
(x) How does bond strength affect IR?
-