Seat No.:	
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NF-130

December-2015

M.Sc., Sem.-III

303 : Chemistry (Organic)

Time: 3 Hours] [Max. Marks: 70

Instructions: (1) All the questions are compulsory.

- (2) Figure to right indicates full marks.
- 1. (A) (1) Write a short note on UV Spectra and polynuclear aromatic hydrocarbons. 4
 - (2) Explain relation between UV Spectra and Geometric isomers. 3

OR

- (1) Explain auxochromes and chromophores. 3
- (2) Calculate the λ max values for the following compounds:

$$(a) \qquad (b) \qquad (b) \qquad (c) \qquad (c) \qquad (d) \qquad (d)$$

- (B) (1) Explain bending vibrations of non-linear AX₂ type molecule in IR Spectra. 4
- (2) Discuss Pallet method to prepare solid samples in IR spectra. 3

OR

- (1) Discuss factors affecting to frequency in IR spectra.
- (2) How can you distinguish styrene and ethyl benzene by IR spectra?

2.	(A)) What is chemical shift? Explain shielded and deshielded protons.		
			OR	
		(1)	Write a note on COSY2D NMR spectroscopy.	4
		(2)	Explain TMS.	3
	(B)	(1)	Write a note on shift reagents in NMR spectra.	4
		(2)	Write a note on nuclear overhauser effect.	3
			OR	
		(1)	Role of fluorine in 'H-NMR spectra.	4
		(2)	How can you distinguish 1-butyne and 2-butyne by 'H NMR spectra?	3
3.	(A)	(1)	Explain Mchafferty rearrangement.	4
		(2)	Give names of various hyphenated techniques of mass spectroscopy.	3
			OR	
		(1)	Give fragmentation of acetophenone.	3
		(2)	Write a note on any two ionization techniques in mass spectroscopy.	4
	(B)	(1)	What is off-resonance decoupling in 13 _C -NMR ?	4
		(2)	Explain how factor like electro-negativity affects the 13 _C chemical shift by	
			taking suitable example.	3
			OR	
		(1)	Explain DEPT spectrum in ¹³ C-NMR.	4
		(2)	Write a short note on ¹³ C-NMR applications.	3
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4. (A) An organic compound with MF $C_7H_{14}O$ exhibits the following spectral data:

IR: 2990, 1690 cm⁻¹.

'H NMR : δ0·9 (6 H, triplet), 1·6 (4H, sextet), 2-4 (4H, triplet)

¹³C NMR : δ 13·6, 16·9, 44·8, 210·8 ppm

HRMS: 114·1855

Deduce the structure of the compound with suitable explanation.

OR

An organic compound with MF $C_{10}H_{12}O_2$ exhibits the following spectral data : 7

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UV: λmax 263 nm 410E

IR: 2832, 1712, 1632, 1598, 1550, 1355, 1328, 1237, 1105, 732, 694 cm⁻¹.

'H NMR : δ 1·12 (6H, doublet), 3·46 (1H, multiplet) 7·21 (5H, singlet)

¹³C NMR : δ 15·1, 27·2, 124·9, 128·2, 135·6, 149·3, 173 ppm

Deduce the structure of the compound with suitable explanation.

(B) An organic compound with a MF $C_5H_8O_2$ exhibits the following spectral data : 7

IR: 1724, 1150 cm⁻¹.

'H NMR : δ 2.7 (4H,t), 4.1 (4H,t)

¹³C NMR : δ 42.9, 68.8, 208.6 ppm

HRMS: 100·1156

Deduce the structure of the compound with suitable explanation.

OR

An organic compound with MF $C_9H_{16}O_4$ exhibits the following spectra data: 7

UV : No strong band

IR (KBr): 2900, 2865, 1735, 1240 cm⁻¹.

'H NMR : δ 1·47 (6H,S), 1·29 (6H,t), 4·14 (4H,q)

¹³C NMR : δ 14·2, 22·8, 50·5, 61·6, 171·4

HRMS: 188·1049.

Deduce the structure of the compound with suitable explanation.

Ansv	ver the following questions in brief:
(1)	Give examples of two solvents used in UV spectra.
(2)	The rules to calculate λmax of diene and enone compounds are called ?
(3)	What is blue shift in UV spectra?
(4)	Why IR spectra called band spectra?
(5)	Give examples of two compounds which are IR transparent.
(6)	What is the use of IR spectra?
(7)	Name any three types of NMR spectra.
(8)	How many signals can Anthracene gives in 'H NMR?
(9)	What is the relation between $\delta(delta)$ and $\tau(Tau)$?
(10)	What is the natural abundance of 13 C?
(11)	How many peaks 3-pentanone give in ¹³ C NMR ?
(12)	What is CSCM?
(13)	What is INADEQUATE?
(14)	Give symbol of radical cation.

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5.

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